Science at the Hart School

Yr 10 Combined Trilogy Curriculum overview

Curriculum intent: Science encompasses everything that we are and allows us to make sense of the world around us. Science at The Hart School is more than just a core subject. We believe an outstanding science education should develop students' curiosity and scientific knowledge to guestion the world in which we live, enable critical-thinking and encourage students to become socially aware global citizens.

Our Science faculty has planned an inspiring, inclusive, and diverse curriculum that is designed to engage and enthuse students with the real-life applications of the subject whilst promoting ambition and aspirations for their future.

In an ever-changing world, in which STEAM subjects are at the forefront of advancements for the future, we want to prepare our students for this by not only looking at the knowledge of the subject, but also the methods, processing skills and applications associated with it. This ensures that our students are scientifically literate, able to evaluate what they see in the news and the world around them and make informed decisions that will affect their future lives and the planet.

	Autumn							Spring		Summer				
Core Course Topic: These topics are taught in	B1: Cell biology	C1: Periodic table & atomic structure	P1: Energy	B2: Organisation	C2: Bonding & properties	C3: Quantitative chemistry	P2: Electricity	C4: Chemical changes		B3: Infection & response	P3: Particle model	B4: Bioenergetics	P4: Atomic structure & radiation	C5: Energetics
Additional	AQA B1 support	- AQA C1 support	AQA P1 support	AQA B2 support	AQA C2 support	AQA C3 support	AQA P2 support	AQA C4 support		AQA B3 support	- AQA P3 support -	AQA B4 support	- AQA P4 support	- AQA C5 support
support links: Here are	BBC bitesize	- BBC bitesize	<u>BBC bitesize</u>	BBC bitesize	- BBC bitesize	- BBC bitesize	<u>BBC bitesize</u>	- BBC bitesize		BBC bitesize	<u>BBC bitesize</u>	<u>BBC bitesize</u>	<u>BBC bitesize</u>	- BBC bitesize
	<u>B1 support video</u> playlist	<u>C1 support</u> video playlist	<u>P1 support video</u> playlist	<u>B2 support video</u> playlist	<u>C2 support</u> video playlist	<u>C3 support</u> video playlist	P2 support video playlist	<u>C4 support</u> <u>video playlist</u>	ent 1	<u>B3 support video</u> playlist	P3 support video playlist	<u>B4 support video</u> playlist	P4 support video playlist	<u>C5 support</u> <u>video playlist</u>
Knowledg e: Included here is the specific knowledg e your child will learn in detail	Cell structure, eukaryotes and prokaryotes, cell specialisation, cel differentiation, sten cells, cell division and the cell cycle, diffusion, active transport, osmosis	Atoms, elements, compounds, mixtures, II seaprating techniques, development of the atomic model, sub- atomic particles, electron configurations, patterns in properties and reactivity in the	Energy stores and systems, energy changes in a system, power, conservation and disspation of energy in a system, efficiency, national and global energy resources	y stores and Animal tissues, is, energy organs and yes in a systems, the i, power, digestive system, rvation and thion of blood vessels, y in a coronary heart i, efficiency, disease, health, ial and risk factors, il energy cancer, plant rces tissues and organ	Ionic, covalent and metallic bonding, states of matter, properties of bonding, allotropes of carbon	Conservation of mass, Relative formula mass, changes in mass when reactant or product is a gas, moles, balancing equations, limiting reactants, concentrations of solutions	Current, potential difference, resistance, resistors, series and parallel , domestic electricity, power, the national grid	Reactivity of metals, extraction of metals and reduction, reactions of acids, pH scale and neutralisation , electrolysis		Communicable diease and pathogens, human defence systems, vaccination, development of drugs	Changes of state, the particle model, density, internal energy, energy transfers, temperature changes in a system, specific heat capacity, pressure in gases	Photosynthesis, uses of glucose, areobic and anaerobic respiration, response to escercise, metabolism	Structure of an atom, isotopes, devlopment of the model of the atom, radioactive decay, nuclear radiation, nuclear equations, contamination	Exothermic and endothermic reactions, reaction profiles, calculating enthalpy changes,
Skills: Included here is the specific skills your child will learn in detail	Required practica 1: Use a light microscope to observe, draw and label a selection of plant and animal cells. A magnification scale must be included.	Develop an understanding of why and how scientific methods and theories change over time.	Required practical activity 14: An investigation to determine the specific heat capacity of one or more materials.	Required practica 3: Use qualitative reagents to test for a range of carbohydrates, lipids and proteins.	Visualise and represent 2D and 3D forms including twodimensional representations of 3D objects.	Substitute numerical values into algebraic equations using appropriate units for physical quantities. Use ratios, fractions and percentages. Change the subject of an equation. Recognise and use	Required practical activity 15: Use circuit diagrams to set up and check appropriate circuits to investigate the factors affecting the resistance of electrical circuits.	Required practical activity 8: Preparation of a pure, dry sample of a soluble salt from an insoluble oxide or carbonate, using a Bunsen burner to heat dilute acid and a water bath or electric heater to evaporate the solution.	Assessm	Understand that the results of testing and trials are published only after scrutiny by peer review.	Required practical activity 17: Use appropriate apparatus to make and record the measurements needed to determine the densities of regular and irregular solid objects and liquids.	Required practica activity 5: Investigate the effect of light intensity on the rate of photosynthesis using an aquatic organism such as pondweed.	Students should be able to recognise expressions given in standard form.	Required practical activity 10: Investigate the variables that affect temperature changes in reacting solutions such as, eg acid plus metals, acid plus carbonates, neutralisations, displacement of metals.
	Use prefixes centi, milli, micro and nano. Re-arrange and use, the magnification equation to calculate magnification. Convert units between milli, micro and nano.	Safe use of a range of , equipment to separate chemical mixtures.	Recall and re- arrange equations to calculate energy changes in a system. Convert between units and express values in standard form.	Required practica 4: Investigate the effect of pH on the rate of reaction of amylase enzyme.			Required practical activity 16: Use circuit diagrams to construct appropriate circuits to investigate the I–V characteristics of a variety of circuit elements,					Investigations into the effect of exercise on the body.	Develop an understanding of why and how scientific methods and theories change over time.	
learning online							https:	//www.caro	usel-learning	a.com/				



